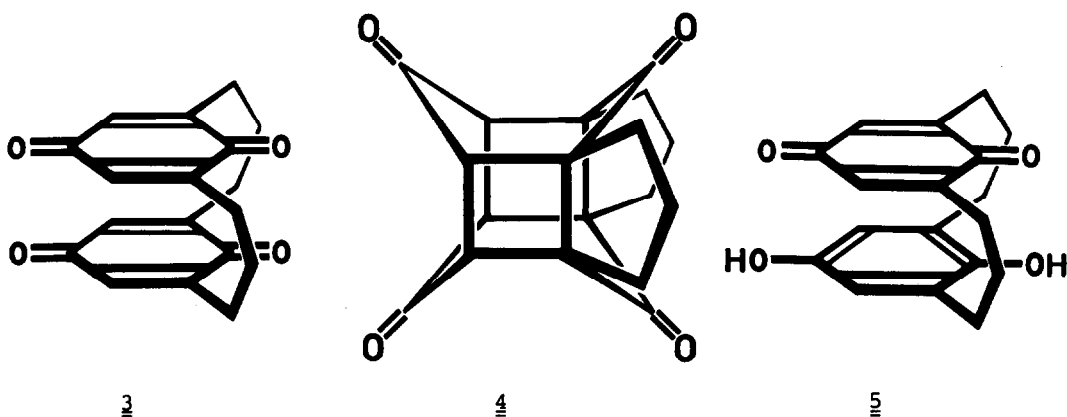




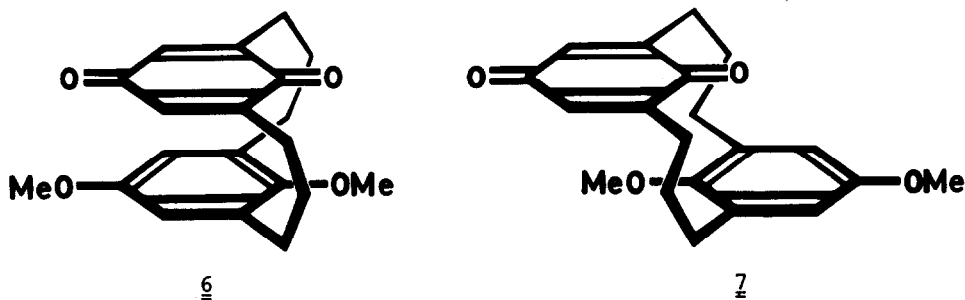
From 1 as well as 2 by demethylation (methylmagnesium iodide, 30 min, 180° C) and subsequent oxidation (silver oxide, acetone) the same [3.3](2,6)-p-benzoquinonophane (3)<sup>3)</sup> was obtained (yellow needles, m.p. 236° C, dec.; 60 % yield). <sup>1</sup>H-NMR data [ $\delta$  = 1.5 - 3.2 (m, 12 H), 6.38 (s, 4 H), CDCl<sub>3</sub>] and the photochemical conversion into the 'cage' isomer 4<sup>3)</sup> [dec. > 280° C; <sup>1</sup>H-NMR:  $\delta$  = 1.6 - 2.3 (m, 12 H), 3.33 (s, 4 H)] suggest the syn constitution for 3. The loss in sterical stability when going from 1/2 to 3 is explained as resulting from smaller size of intraanular substituents and higher flexibility of quinoid rings. Catalytic hydrogenation of 3 (Pd/C, dioxane, 20° C) yielded the quinhydrone 5<sup>3)</sup> [dark-violet crystals, m.p. 245 - 246° C (dec.); 82 % yield] the CT absorption of which [ $\lambda_{\max}$  447 nm ( $\epsilon$  2740), in dioxane] is very similar to the CT absorption of pseudogeminal [3.3]paracyclophane quinhydrone <sup>1)</sup> suggesting an analogous donor-acceptor orientation.



In an attempt to obtain quinhydrones both of the syn- and of the anti-[3.3]metacyclophane series the partial demethylations of 1 and 2 were studied. Treatment of 1 with chlorotrimethylsilane/sodium iodide in acetonitril <sup>4)</sup> (2 d, 40° C) after oxidation with silver oxide led in 38 % yield to syn-15,18-dimethoxy[3](2,6)-p-benzoquinono[3]metacyclophane (6)<sup>3)</sup> [red prisms, m.p. 195 - 197° C; <sup>1</sup>H-NMR (CDCl<sub>3</sub>):  $\delta$  = 1.7 - 3.1 (m, 12 H), 3.58 (s, 3 H), 3.63 (s, 3 H), 5.96 (s, 2 H), 6.35 (s, 2 H)]. The strong CT absorption of 6 [ $\lambda_{\max}$  400 nm ( $\epsilon$  2740), in chloroform] supported again the assumption of a close resemblance of mutual donor-acceptor orientations in syn-[3.3]metacyclophane

and [3.3]paracyclophane quinhydrones which definitely was confirmed by X-ray structure analysis of 6<sup>5</sup>): in the projection of the quinone ring onto the benzene ring the ring centers show a transannular distance of 3.19 Å and are only slightly dislocated; the inclination of the C(4)-C(5)-C(7)-C(8) and C(13)-C(14)-C(16)-C(17) planes is only 15°.

As a by-product in the 6-synthesis a monomethyl ether of 1 has been obtained (red crystals, dec. > 184° C; 11.5 % yield) the CT absorption of which  $\lambda_{\max}$  438 nm ( $\epsilon$  2610), in dioxane] is nearly congruent with that of 1.



Preliminary experiments on partial demethylation of 2 (methylmagnesium iodide, 160° C) led, after oxidation with silver oxide, to an isomer of 6 of m.p. 168 - 169° C ( $M^+$ : calc. 326.1518, obs. 326.1522). <sup>1</sup>H-NMR data [ $\delta$  = 2.2 - 3.2 (m, 12 H), 3.12 (s, 3 H), 3.69 (s, 3 H), 6.12 (s, 2 H), 6.40 (s, 2 H), in CDCl<sub>3</sub>] agree well with the structure of the anti-15,18-dimethoxy[3](2,6)-p-benzoquinono[3]metacyclophane (7). However, alternative structures (e.g., a quinol isomer of 7) have not yet been ruled out definitely. The compound obtained does not at all show a CT absorption extending beyond 380 nm. Thus, in case we are really dealing with 7, the pair of stereoisomers 6/7 would offer an especially striking demonstration of the orientation dependence of CT absorptions.

- 1) W. Rebafka and H. A. Staab, Angew. Chem. 85, 831 (1973); 86, 234 (1974); Angew. Chem. Int. Ed. Engl. 12, 776 (1973); 13, 203 (1974); H. A. Staab and W. Rebafka, Chem. Ber. 110, 3333 (1977); H. A. Staab and C. P. Herz, Angew. Chem. 89, 839 (1977); Angew. Chem. Int. Ed. Engl. 16, 799 (1977); R. Reimann and H. A. Staab, *ibid.* 90, 385 (1978), and 17, 374 (1978), *resp.*; H. A. Staab and V. Schwendemann, *ibid.* 90, 805 (1978), and 17, 756 (1978), *resp.*; H. A. Staab and U. Zapf, *ibid.* 90, 807 (1978), and 17, 757 (1978), *resp.*; further references are given there.
- 2) H. A. Staab, C. P. Herz and A. Döhling, Chem. Ber. (in press).
- 3) Elemental analyses, mass spectra, and other spectroscopic properties are in agreement with the structures given.
- 4) Cf. T. Morita, Y. Okamoto and H. Sakurai, J. C. S. Chem. Comm. 1978, 874.
- 5) C. Krieger and H. A. Staab, to be published.

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