## [3.3] METACYCLOPHANE QUINHYDRONES

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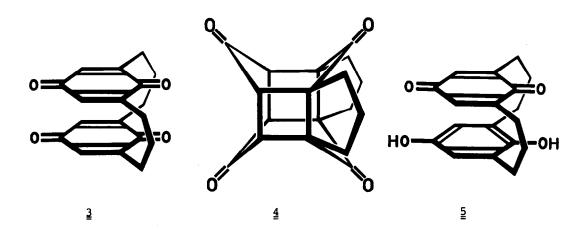
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Via the stable stereoisomers  $\underline{\underline{l}}$  and  $\underline{\underline{l}}$  of 6,9,15,18-tetramethoxy[3.3]metacyclophane, by demethylation and oxidation, intramolecular quinhydrones of the [3.3]metacyclophane series were obtained the charge-transfer absorption of which is discussed.

Orientation and distance dependence of charge-transfer (CT) absorptions has been studied extensively by investigations on intramolecular donor-acceptor systems with well-defined geometry <sup>1)</sup>. In this connection we tried to determine the effect of a parallel translation of donor and acceptor units by comparing syn- and anti-[3.3] metacyclophane quinhydrones.

The two stereoisomeric 7,10,17,20-tetramethoxy-2,13-dithia[4.4] metacyclophanes  $^{2)}$ , by pyrolysis (570° C,  $10^{-4}$  Torr) of the disulfones derived therefrom, yielded <u>syn-</u> and <u>anti-6,9,15,18-tetramethoxy[3.3] metacyclophanes</u> which were separated by chromatography:  $\frac{1}{2}$  (m.p.  $115-116^{\circ}$  C, 26 % yield) and  $\frac{2}{2}$  (m.p.  $89.5^{\circ}$  C, 21 % yield) the assignment of which is proved by  $\frac{1}{1}$ H-NMR [ $\frac{1}{2}$ :  $\delta = 1.6-3.4$  (m, 12 H), 3.46 (s, 6 H), 3.67 (s, 6 H), 6.12 (s, 4 H);  $\frac{2}{2}$ :  $\delta = 1.9-2.9$  (m, 12 H), 3.14 (s, 6 H), 3.78 (s, 6 H), 6.54 (s, 4 H), CDC1 $_3$ ].  $\frac{1}{2}$  and  $\frac{2}{2}$  are remarkably stable stereoisomers; no isomerisation was detectable even after prolonged heating (e.g., 25 h at  $190^{\circ}$  C in decalin).

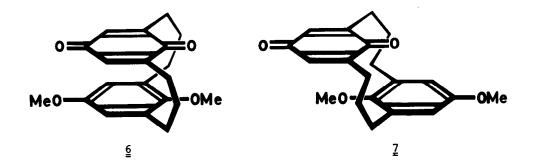
From  $\frac{1}{2}$  as well as  $\frac{2}{2}$  by demethylation (methylmagnesium iodide, 30 min,  $180^{\circ}$  C) and subsequent oxidation (silver oxide, acetone) the same [3.3](2,6)-p-benzoquinonophane ( $\frac{3}{2}$ ) was obtained (yellow needles, m.p.  $236^{\circ}$  C, dec.; 60 % yield).  $^{1}$ H-NMR data [6 = 1.5-3.2 (m, 12 H), 6.38 (s, 4 H), CDCl $_{3}$ ] and the photochemical conversion into the 'cage' isomer  $\frac{4}{2}$  [dec. >  $280^{\circ}$  C;  $^{1}$ H-NMR: 6 = 1.6-2.3 (m, 12 H), 3.33 (s, 4 H)] suggest the  $\frac{1}{2}$  constitution for  $\frac{3}{2}$ . The loss in sterical stability when going from  $\frac{1}{2}$  to  $\frac{3}{2}$  is explained as resulting from smaller size of intraanular substituents and higher flexibility of quinoid rings. Catalytic hydrogenation of  $\frac{3}{2}$  (Pd/C, dioxane,  $20^{\circ}$  C) yielded the quinhydrone  $\frac{5}{2}$  [dark-violet crystals, m.p.  $245-246^{\circ}$  C (dec.); 82 % yield] the CT absorption of which  $\frac{1}{2}$   $\frac{1}{2}$   $\frac{1}{2}$  to  $\frac{1}{2}$  in dioxane] is very similar to the CT absorption of pseudogeminal [3.3] paracyclophane quinhydrone  $\frac{1}{2}$  suggesting an analogous donor-acceptor orientation.



In an attempt to obtain quinhydrones both of the syn- and of the anti- [3.3]metacyclophane series the partial demethylations of  $\frac{1}{2}$  and  $\frac{2}{2}$  were studied. Treatment of  $\frac{1}{2}$  with chlorotrimethylsilane/sodium iodide in acetonitril 4) (2 d, 40° C) after oxidation with silver oxide led in 38 % yield to syn- 15,18-dimethoxy[3](2,6)-p-benzoquinono[3]metacyclophane ( $\frac{6}{2}$ ) [red prisms, m.p. 195-197° C;  $\frac{1}{2}$ H-NMR (CDCl<sub>3</sub>):  $\delta$  = 1.7 - 3.1 (m, 12 H), 3.58 (s, 3 H), 3.63 (s, 3 H), 5.96 (s, 2 H), 6.35 (s, 2 H)]. The strong CT absorption of  $\frac{6}{2}$  [ $\lambda$ <sub>max</sub> 400 nm ( $\epsilon$  2740), in chloroform] supported again the assumption of a close resemblance of mutual donor-acceptor orientations in syn-[3.3]metacyclophane

and [3.3]paracyclophane quinhydrones which definitely was confirmed by X-ray structure analysis of  $\underline{6}^{5}$ : in the projection of the quinone ring onto the benzene ring the ring centers show a transanular distance of 3.19  $\overset{\text{O}}{\text{A}}$  and are only slightly dislocated; the inclination of the C(4)-C(5)-C(7)-C(8) and C(13)-C(14)-C(16)-C(17) planes is only  $15^{\circ}$ .

As a by-product in the  $\underline{6}$ -synthesis a monomethyl ether of  $\underline{1}$  has been obtained (red crystals, dec. >  $184^{\circ}$  C; 11.5 % yield) the CT absorption of which  $\Omega_{\text{max}}$  438 nm (£2610), in dioxane) is nearly congruent with that of  $\underline{1}$ .



Preliminary experiments on partial demethylation of  $\underline{2}$  (methylmagnesium iodide,  $160^{\circ}$  C) led, after oxidation with silver oxide, to an isomer of  $\underline{6}$  of m.p.  $168-169^{\circ}$  C (M<sup>+</sup>: calc. 326.1518, obs. 326.1522). <sup>1</sup>H-NMR data [ $\delta$  = 2.2 - 3.2 (m, 12 H), 3.12 (s, 3 H), 3.69 (s, 3 H), 6.12 (s, 2 H), 6.40 (s, 2 H), in CDCl<sub>3</sub>] agree well with the structure of the anti-15,18-dimethoxy[3](2,6)-p-benzoquinono[3]metacyclophane ( $\underline{7}$ ). However, alternative structures (e.g., a quinol isomer of  $\underline{7}$ ) have not yet been ruled out definitely. The compound obtained does not at all show a CT absorption extending beyond 380 nm. Thus, in case we are really dealing with  $\underline{7}$ , the pair of stereoisomers  $\underline{6}/\underline{7}$  would offer an especially striking demonstration of the orientation dependence of CT absorptions.

- 1) W. Rebafka and H. A. Staab, <u>Angew. Chem.</u> §5, 831 (1973); §6, 234 (1974); <u>Angew. Chem. Int. Ed. Engl.</u> §2, 776 (1973); §3, 203 (1974); H. A. Staab and W. Rebafka, <u>Chem. Ber.</u> §2, 3333 (1977); H. A. Staab and C. P. Herz, <u>Angew. Chem.</u> §9, 839 (1977); <u>Angew. Chem. Int. Ed. Engl.</u> §6, 799 (1977); R. Reimann and H. A. Staab, ibid. §9, 385 (1978), and §7, 374 (1978), resp.; H. A. Staab and V. Schwendemann, ibid. §9, 805 (1978), and §7, 756 (1978), resp.; H. A. Staab and U. Zapf, ibid. §9, 807 (1978), and §7, 757 (1978), resp.; further references are given there.
- 2) H. A. Staab, C. P. Herz and A. Döhling, Chem. Ber. (in press).
- 3) Elemental analyses, mass spectra, and other spectroscopic properties are in agreement with the structures given.
- 4) Cf. T. Morita, Y. Okamoto and H. Sakurai, J. C. S. Chem. Comm. 1978, 874.
- 5) C. Krieger and H. A. Staab, to be published.
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